| Question |  |  | Answers | Notes | Total |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 2. | a | i | read off between 17 and 19 «deg» $\checkmark$ correct use of $d=\frac{\lambda}{\sin \theta}=7.8 \times 10^{-15} « \mathrm{~m} » \checkmark$ $\text { so radius }=\frac{7.8}{2} \text { «fm» }=3.9 \text { «fm» } \checkmark$ | Award ecf for wrong angle in MP1. <br> Answer for MP3 must show at least 2 sf. | 3 |
| 2. | a | ii | $R_{\mathrm{Th}}=R_{\mathrm{Si}}\left(\frac{A_{\mathrm{Th}}}{A_{\mathrm{si}}}\right)^{\frac{1}{3}}$ or substitution $\checkmark$ $7.4 \text { «fm» }$ |  | 2 |
| 2. | a | iii | electron wavelength shorter than alpha particles (thus increased resolution) OR electron is not subject to strong nuclear force $\checkmark$ |  | 1 |
| 2. | a | iv | nuclear forces act $\checkmark$ <br> nuclear recoil occurs $\checkmark$ <br> significant penetration into nucleus / probing internal structure of individual nucleons $\sqrt{ }$ <br> incident particles are relativistic $\checkmark$ |  | 2 max |

(Question 2 continued)

| Question |  |  | Answers | Notes | Total |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 2. | b | i | $\begin{aligned} & { }_{15}^{30} \mathrm{P} \rightarrow\left({ }_{14}^{30} \mathrm{Si}\right) \checkmark \\ & +{ }_{+1}^{0} \mathrm{e}+v_{\mathrm{e}} \checkmark \end{aligned}$ |  | 2 |
| 2. | b | ii | correct change of either $u$ to $d \checkmark$ <br> W+ shown $\checkmark$ <br> correct arrow directions for positron and electron neutrino $\checkmark$ |  | 3 |
| 2. | b | iii | quarks cannot be directly observed as free particles/must remain bound to other quarks/quarks cannot be isolated $\checkmark$ because energy given to nucleon creates other particles rather than freeing quarks/OWTTE $\checkmark$ |  | 2 |

(Question 2 continued)

| Question |  | Answers | Notes | Total |
| :--- | :--- | :--- | :--- | :--- | :---: |
| 2. | c | models need testing/new information may change models/new technology <br> may bring new information/Models can be revised/OWTTE $\checkmark$ | $\mathbf{1}$ |  |


(Question 8 continued)

| Question |  |  | Answers | Notes | Total |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 8. | b | i | with $n=3, v=« \sqrt{\frac{2 \times 8.99 \times 10^{9} \times\left(1.6 \times 10^{-19}\right)^{2}}{9.11 \times 10^{-31} \times 9 \times 2.7 \times 10^{-11}}}=» 1.44 \times 10^{6}<\mathrm{ms}^{-1}$ » $\lambda=\frac{6.63 \times 10^{-34}}{9.11 \times 10^{-31} \times 1.44 \times 10^{6}} \text { OR } \lambda=5.05 \times 10^{-10} « \mathrm{~m} » \downarrow$ |  | 2 |
| 8. | b | ii | $\frac{2 \pi r}{\lambda}=« \frac{2 \pi \times 9 \times 2.7 \times 10^{-11}}{5.1 \times 10^{-10}}=2.99 » \cong 3$ ل | Allow ECF from (b)(i) | 1 |
| 8. | c |  | reference to fixed orbits/specific radii $O \boldsymbol{R}$ quantized angular momentum in Bohr model $\sqrt{ }$ electron described by a wavefunction/as a wave in Schrödinger model $O R$ as particle in Bohr model $\sqrt{ }$ <br> reference to «same» energy levels in both models $\checkmark$ <br> reference to «relationship between wavefunction and» probability «of finding an electron in a point» in Schrödinger model $\checkmark$ |  | 3 max |


| Question |  |  | Answers | Notes | Total |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 11. | a | i | «low intensity light would» transfer energy to the electron at a low rate/slowly $\checkmark$ time would be required for the electron «to absorb the required energy» to escape/be emitted $\checkmark$ | OWTTE | 2 |
| 11. | a | ii | «in the photon theory of light» the electron interacts with a single photon $\boldsymbol{\checkmark}$ and absorbs all the energy $O \boldsymbol{R}$ and can leave the metal immediately $\checkmark$ | Reference to photon-electron collision scores MP1 | 2 |
| 11. | b | i | $\begin{aligned} & \phi=\frac{h c}{\lambda}-E_{\mathrm{K}} \checkmark \\ & E_{\mathrm{K}}=1.5 « \mathrm{eV} » \checkmark \\ & \phi=« \frac{1.24 \times 10^{-6}}{480 \times 10^{-9}}-1.5=» 1.1 « \mathrm{eV} » \quad \end{aligned}$ | Allow reading from the graph of $E_{k}=1.4$ leading to an answer of 1.2 «eV». | 3 |

(continued...)
(Question 11 continued)

| Question |  |  | Answers | Notes | Total |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 11. | b | ii | similar curve lower than original $\checkmark$ with same horizontal intercept $\checkmark$ |  | 2 |


| Question |  |  | Answers | Notes | Total |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 11. | a |  | «de Broglie's hypothesis states that the» electron is represented by a wave $\checkmark$ therefore it cannot be localized/it is spread out/it does not have a definite position $\checkmark$ | Award MP1 for any mention of wavelike property of an electron. | 2 |
| 11. | b | i | $\begin{aligned} & « d \sin \theta=\lambda \Rightarrow » d=\frac{1.6 \times 10^{-15}}{\sin 17^{\circ}} / 5.47 \times 10^{-15} \text { «m» } \\ & R=\frac{d}{2} \approx 2.7 / 2.8 \times 10^{-15} « \mathrm{~m} » \end{aligned}$ |  | 2 |
| 11. | b | ii | this implies that the nucleons are very tightly packed/that there is very little space in between the nucleons $\checkmark$ <br> because the nuclear force is stronger than the electrostatic force $\checkmark$ |  | 2 |
| 11. | c | i | number of nuclei is $\frac{28 \times 10^{-3}}{64} \times 6.02 \times 10^{23} / 2.63 \times 10^{20} \checkmark$ $A=« \lambda N=2.63 \times 10^{20} \times \frac{5.5 \times 10^{-2}}{3600} »=4.0 \times 10^{15} \text { «Bq» } \checkmark$ |  | 2 |
| 11. | c | ii | $\begin{aligned} & \frac{1}{3}=e^{-\lambda t} \checkmark \\ & t=20 « h r » \end{aligned}$ |  | 2 |


| 8. | a |  | $E_{1}=-13.6 « \mathrm{eV} » E_{2}=-\frac{13.6}{4}=-3.4 « \mathrm{eV} » \checkmark$ <br> energy of photon is difference $E_{2}-E_{1}=10.2 « \approx 10 \mathrm{eV}$ » $\checkmark$ | Must see at least 10.2 eV . | 2 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 8. | b | i | $\begin{aligned} & 10-5.1=4.9 \text { «eV» } \\ & 4.9 \times 1.6 \times 10^{-19}=7.8 \times 10^{-19} \text { «J» } \end{aligned}$ | Allow 5.1 if 10.2 is used to give $8.2 \times 10^{-19}$ «J». | 2 |
| 8. | b | ii | EPE produced by battery $\checkmark$ exceeds maximum KE of electrons / electrons don't have enough KE $\checkmark$ | For first mark, accept explanation in terms of electric potential energy difference of electrons between surface and plate. | 2 |
| 8. | b | iii | 4.9 «V» | Allow 5.1 if 10.2 is used in (b)(i). Ignore sign on answer. | 1 |

(continued...)
(Question 8 continued)

| 8. | C | i | two equally spaced vertical lines (judge by eye) at approximately $1 / 3$ and $2 / 3 \checkmark$ labelled correctly $\checkmark$ |  | 2 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 8. | C | ii | kinetic energy at collecting plate $=0.9$ «eV» $\checkmark$ $\text { speed }=« \sqrt{\frac{2 \times 0.9 \times 1.6 \times 10^{-19}}{9.11 \times 10^{-31}}} »=5.6 \times 10^{5}<\mathrm{ms}^{-1} » \checkmark$ | Allow ECF from MP1 | 2 |

(Question 8 continued)

| Question |  | Answers | Total |
| :--- | :--- | :--- | :--- | :--- | :---: |
| 8. | d | OR base of the thundercloud must be parallel to the Earth surface <br> the base of the thundercloud must be flat <br> OR <br> the base of the cloud must be very long «compared with the distance from the <br> surface» $\checkmark$ | $\mathbf{1}$ |


| 9. | $\mathbf{a}$ |  | «most of» the mass of the atom is confined within a very small <br> volume/nucleus $\checkmark$ <br> «all» the positive charge is confined within a very small volume/nucleus $\checkmark$ <br> electrons orbit the nucleus «in circular orbits» $\checkmark$ |  |
| :--- | :--- | :--- | :--- | :--- |
| 9. | $\mathbf{b}$ | the electrons accelerate and so radiate energy $\checkmark$ <br> they would therefore spiral into the nucleus/atoms would be unstable $\checkmark$ <br> electrons have discrete/only certain energy levels $\checkmark$ <br> the only orbits where electrons do not radiate are those that satisfy the Bohr <br> condition «mvr=n $\frac{h}{2 \pi}$ » $\checkmark$ | $\mathbf{3}$ max |  |

(Question 9 continued)

| Question |  |  | Answers | Notes | Total |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 9. | C | i | $\frac{m_{\mathrm{e}} v^{2}}{r}=\frac{k e^{2}}{r^{2}}$ <br> OR <br> $\mathrm{KE}=\frac{1}{2} \mathrm{PE}$ hence $\frac{1}{2} m_{e} v^{2}=\frac{1}{2} \frac{\mathrm{ke}^{2}}{r} \checkmark$ «solving for $v$ to get answer» | Answer given - look for correct working | 1 |
| 9. | c | ii | combining $v=\sqrt{\frac{k e^{2}}{m_{\mathrm{e}} r}}$ with $m_{\mathrm{e}} v r=\frac{h}{2 \pi}$ using correct substitution $\checkmark$ $\text { «eg } m_{\mathrm{e}}{ }^{2} \frac{k e^{2}}{m_{\mathrm{e}} r} r^{2}=\frac{h^{2}}{4 \pi^{2}} »$ <br> correct algebraic manipulation to gain the answer $\checkmark$ | Answer given - look for correct working <br> Do not allow a bald statement of the answer for MP2. Some further working eg cancellation of $m$ or $r$ must be shown | 2 |
| 9. | c | iii | $\begin{aligned} & « r=\frac{\left(6.63 \times 10^{-34}\right)^{2}}{4 \pi^{2} \times 8.99 \times 10^{9} \times 9.11 \times 10^{-31} \times\left(1.6 \times 10^{-19}\right)^{2}} » \\ & r=5.3 \times 10^{-11} « \mathrm{~m} » \checkmark \end{aligned}$ |  | 1 |
| 9. | d | i | the energy released is $3.54-0.48=3.06 « \mathrm{MeV} » \checkmark$ this is shared by the electron and the antineutrino $\checkmark$ so the electron's energy varies from 0 to 3.06 «MeV» $\checkmark$ |  | 3 |
| 9. | d | ii | the palladium nucleus emits the photon when it decays into the ground state «from the excited state» |  | 1 |

(Question 9 continued)

| Question |  |  | Answers | Notes | Total |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 9. | d | iii | Photon energy $\begin{aligned} & \mathrm{E}=0.48 \times 10^{6} \times 1.6 \times 10^{-19}=« 7.68 \times 10^{-14} \mathrm{~J} » \\ & \lambda=« \frac{h c}{E}=\frac{6.63 \times 10^{-34} \times 3 \times 10^{8}}{7.68 \times 10^{-14}}=» 2.6 \times 10^{-12} « \mathrm{~m} » \end{aligned}$ | Award [2] for a bald correct answer <br> Allow ECF from incorrect energy | 2 |
| 9. | e | i | line with arrow as shown labelled anti-neutrino $\bar{v} \checkmark$ | Correct direction of the "arrow" is essential <br> The line drawn must be "upwards" from the vertex in the time direction i.e. above the horizontal <br> eg: | 1 |
| 9. | e | ii | $V=W^{-} \checkmark$ |  | 1 |

